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### **RESEARCH ARTICLE**

## Simultaneous Determination and Validation of Hydrochlorothiazide & Eprosartan in bulk drug and Marketed Formulation by Rp-Hplc

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#### ABSTRACT

Validation of developed and RP HPLC analytical method according to ICH Guidelines. A review of literature reveals that no analytical methods are available for these drugs. There are no methods in the literature for the estimation of drugs in combined dosage forms. Analytical methods have been report for hydrochlorothiazide and eprosartan using RP-HPLC in pharmaceutical dosage form. The development of simple and precise reverse phase high performance liquid chromatography of hydrochlorothizide and eprosartan using single method in bulk in bulk drug and market formulation.

#### Key Words- Hydrochlorothizide, Eprosartan, RP- HPLC

#### **INTRODUCTION**

Chromatography is a non-destructive procedure for resolving a multi-component mixture of trace, minor, or major constituents into its individual fractions. Different variations may be applied to solids, liquids, and gases. While chromatography applied qualitatively may be both and quantitatively, it is primarily a separation tool. Quantitative analysis can be carried out by measuring the area of the chromatographic peak. Chromatography may be defined as a method of separating a mixture of components into individual components through equilibrium distribution between two phases.

# The chromatographic method of separation, in general, involves the following steps:

- Adsorption or retention of a substance or substances on the stationary phase.
- Separation of the adsorbed substances by the mobile phase.
- Recovery of the separated substances by a continuous flow of the mobile phase; the method being called elution.
- Qualitative and quantitative analysis of the eluted substances1.

# PRINCIPLE OF SEPARATION IN HPLC

The principle of separation in normal phase mode and reverse phase mode is adsorption. When mixtures of components are introduced in to a HPLC column, they travel according to their relative affinities towards the stationery phase. The component which has more affinity towards the adsorbent travels slower. The component which has less affinity towards the stationary phase travels faster. Since no two components have the same affinity towards the stationary phase, the components are separated.

The most important component of HPLC in solvent delivery system is the pump, because its performance directly effects the retention time, reproducibility and detector sensitivity. Among the several solvent delivery systems (direct gas pressure, pneumatic intensifier, reciprocating etc.) reciprocating pump with twin or triple pistons is widely used, as this system gives less baseline noise, good flow rate reproducibility etc.

#### MATERIALS

 Table No. 1. Chemicals and Solvents Used

S. No.	Chemicals	Manufacture
01	Eprosartan	Macleod's Pharmaceutical Ltd.
02	Hydrochlorothiazide	Kalindi Medicure Pvt. Ltd.
03	3 Acetonitrile	Merck Ltd., India

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04	Methenol	Merck Ltd., India
05	water	Merck Ltd., India

#### **METHODS FTIR SPECTRUM**

IR absorption spectra Eprosartan and Hydrochlorothiazide was obtained by KBr pellet method.

#### **Physiochemical characteristics**

• Melting point- M.P. of the Eprosartan and Hydrochlorothiazide was determined by melting point apparatus (lab hosp) and found to be 248-250°C and 270-273 °C respectively.

#### ANALYTICAL METHOD DEVELOPMENT OF EPROSARTAN AND HYDROCHLOROTHIAZIDE BY HPLC:-

# I. Mobile Phase Selection:-

Initially to estimate Eprosartan and Hydrochlorothiazide number of mobile phase in different ratio were tried. Results are shown in (Table no 02 & 03.). Taking into consideration the

system suitability parameter like RT, Tailing factor, No. of theoretical plates and HETP, the mobile phase found to be most suitable for analysis was 50 mM KH2PO4 (pH4.0 with OPA): Acetonitrile in the ratio of 30:70 v/v. The mobile phase was filtered through 0.45 filter paper to particulate matter remove and then degassed by sonication. Flow rate employed for analysis was 1.0 ml/min.

#### II. Selection of wavelength:-

100 mg of Eprosartan and Hydrochlorothiazide was weighed accurately and transferred to a 100 ml volumetric flask, and the volume was adjusted to the mark with the mobile phase in the ratio of 30:70 v/v. From above solutions of 0.1 ml was transferred to 10 ml volumetric flasks, and make up the volume up to mark. Resulting solution was scanned over UV range (200-400nm), maximum absorbance was found at  $\lambda$  max 270 nm.



Figure 1. Wavelength of Eprosartan and Hydrochlorothiazide

# III. Preparation of Standard Stock Solution:-

10 mg of Eprosartan and 10 mg of Hydrochlorothiazide was weighed accurately and transferred to separate 10 ml volumetric flask, and the volume was adjusted to the mark with the mobile phase (50 mm KH2PO4 (pH4.0 with OPA): Acetonitrile in the ratio of 30:70 v/v) to give a stock solution of 1000 g/ml.

#### IV. Preparation of Working Standard Solution:-From stock solutions of Eprosartan 1 ml was taken and diluted up to 10 ml from

was taken and diluted up to 10 ml. from this solution 0.8, 1.6, 2.4, 3.2 and 4.0 ml solutions were transferred to 10 ml volumetric flasks and make up the volume up to 10 ml with mobile phase, gives standard drug solution of 8, 16, 24, 32, and 40  $\mu$ g/ ml concentration and From stock solutions of Hydrochlorothiazide 1 ml was taken and diluted up to 10 ml. from this solution 0.5, 1.0, 1.5, 2.0 and 2.5 ml solutions were transferred to 10 ml volumetric flasks and make up the volume up to 10 ml with mobile phase, gives standard drug solution of 5, 10, 15, 20 and 25  $\mu$ g/ ml concentration.

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Figure - Calibration Graph of Hydrochlorothiazide

## V. Assay of tablet formulation

For analysis of the formulation, twenty tablets taken and determine the average









#### **SOLUBILITY:-**

Table No.2 - Solubility of Eprosartan mesylate

S.No.	Solvent	Solubility
01	Water	Insoluble
02	0.1N HCl	Insoluble
03	Methenol	Freely soluble
04	Ethenol	Freely soluble
05	Acetone	Freely soluble
06	Methenol:water (80:20)	Springly soluble

#### Table No. 3 -Solubility of Hydrochlorothiazide

S.No.	Solvent	Solubility           Insoluble		
01	Water			
02	0.1N HCl	Insoluble		
03	Methenol	Freely soluble		
04	Ethenol	soluble		
05	Acetone	soluble		
06	Methenol:water (80:20)	soluble		

# Linearity and Calibration Graph:-

 Table No.4 - Result of Linearity of Eprosartan

Std. Conc.	0	8	16	24	32	40
(µg/ml)						
1	0	5214.365	10454.236	14689.125	20578.365	25455.236
2	0	5245.145	10467.336	14685.256	20547.235	25469.365
3	0	5241.363	10468.225	14701.365	20569.325	25471.236
Mean	0	5233.624	10463.27	14691.92	20564.98	25465.28
SD	0.00	13.70565	6.395245	6.866073	13.07571	7.142434
%RSD	0.000	0.261877	0.061121	0.046734	0.063582	0.028048



Figure - Calibration Graph of Eprosartan

#### **Regression Equation**

Y = mx + c,

Y= AUC m= slope =634.1 X= Conc. in g/ml c= Intercept = 54.43  $r_2 = 0.999$ 

Std. Conc. (µg/ml)	0	5	10	15	20	25
1	0	3257.359	6574.236	9245.365	12458.365	15478.369
2	0	3267.89	6547.236	9285.658	12468.365	15489.365
3	0	3285.589	6589.236	9241.369	12425.365	15539.659
Mean	0	3270.279	6570.236	9257.464	12450.7	15502.46
SD	0.00	11.64803	17.37815	20.0028	18.37269	26.68096
%RSD	0.000	0.356178	0.264498	0.216072	0.147563	0.172108

Table No.5 - Result for Linearity of Hydrochlorothiazide



Figure 9- Calibration Graph of Hydrochlorothiazide

#### **Regression Equation**

Y= mx +c, m= slope = 384.7

#### **CONCLUSION-**

The method can be used for the estimation of Eprosartan and Hydrochlorothiazide simultaneously from its Tablet dosage form shows the Recovery results of the two different formulations on addition of Standard drug by HPLC. The Mean percentage Recovery and RSD were found to be less than 2 so the method was found to be more accurate without any interference.

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X= Conc. in  $\Box$ g/ml c= Intercept 146.0 r2 = 0.999

TableNo.6-ResultofAnalysisforEprosartanandHydrochlorothiazide in Tablet Formulation

µg/ml	EPROSARTAN	HYDROCHLOROT HIAZIDE	
	32	12.5	
Rep-1	31.99	12.5	
Rep-2	31.99	12.45	
Rep-3	32.00	12.48	
% found*			
Rep-1	99.96875	100	
Rep-2	100	99.6	
Rep-3	100.0313	100.241	
SD	0.31255	0.323754	
%RSD	0.031255	0.323925	

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